# A Synthesis of All Stereoisomers of Tenuecyclamide A Employing a Fluorous-Fmoc Strategy

Yuya Sugiyama, Fumitaka Eguchi, Atsushi Miyazaki, Koichi Hayashi, Hiroaki Takahashi, Hiromi Hamamoto, Takayuki Shioiri, and Masato Matsugi\*

Department of Applied Biological Chemistry, Faculty of Agriculture, Meij[o](#page-7-0) University, 1-501 Shiogamaguchi, Tempaku-ku, Nagoya 468-8502, Japan

#### **S** Supporting Information

[AB](#page-7-0)STRACT: [A concise liq](#page-7-0)uid-phase combinatorial synthesis of all stereoisomers of Tenuecyclamide A was achieved using a mixture of D-/L-alanine with each stereoisomer encoded by a different f-Fmoc tag. The synthetic strategy using f-Fmoc reagents as the protecting group for amino acids has been demonstrated to be a useful method for diverse polypeptide analogue synthesis.



# ■ INTRODUCTION

Since its introduction in 1997 by Curran et al., fluorous solidphase extraction  $(FSPE)^1$  has proven to be a highly efficient and widely applicable technique for separating molecules that are tagged with a fluor[ou](#page-7-0)s domain from other nonfluorous organic molecules.<sup>2</sup> In recent years, the technique has evolved such that molecules can be separated purely on the basis of respective fluorin[e](#page-7-0) content regardless of other functionality present in the remainder of the molecule.<sup>3</sup> This salient feature of FSPE allows for liquid-phase combinatorial synthesis called fluorous mixture synthesis (FMS). In [t](#page-7-0)his technique, the fluorous tags containing differing amounts of fluorine are installed onto the various starting materials, which are then mixed for the remainder of the synthesis and separated based purely on fluorine content at the end, thereby greatly reducing the number of synthetic steps required to build up a compound library.<sup>4</sup>

A particularly novel application of FMS resides in deter[min](#page-7-0)ing the stereochemical requirements of a molecule for it to impart its biological activity. More specifically, the biological activity of peptides is governed by the configuration at the asymmetric center.<sup>5</sup> From a drug discovery perspective, it is interesting to study the biological activity of the non-natural configuration to ascerta[in](#page-7-0) whether or not this modification enhances or diminishes the biological activity or whether or not it elicits an entirely different response. Synthetically, this becomes a daunting task especially in the case of poly peptides where for each additional center there are  $2<sup>n</sup>$  possible stereoisomers. FMS, however, is uniquely suited to address this challenge because the various isomers can be individually labeled with different length fluorous tags then mixed together. The mixture can then be carried through the synthesis and then separated at the very end purely based on the fluorine content, thereby deconvoluting the entire mixture. This strategy saves a significant amount of effort because the synthetic steps between the mixing and the demixing do not have to be carried out on

each individual isomer yet in the end does provide the entire family of desired stereoisomers isolated individually.

In peptide synthesis, the Fmoc reagent is one of the most well-known nitrogen protecting groups for amino acids.<sup>6</sup> We have previously reported a synthetic route toward two kinds of fluorous-Fmoc reagents  $(f\text{-Fmoc})$  and demonstrated that fluorous tagged dipeptides can be separated due to the remarkable differences in their ret[en](#page-7-0)tion times on a fluorous HPLC (f-HPLC) column.<sup>8</sup> Based on the success of this strategy for dipeptides, we planned to use different f-Fmoc reagents with varying amounts of fluori[ne](#page-7-0) as both a nitrogen protecting group and a fluorous encoding tag for a liquid-phase combinatorial peptide synthesis. Herein, we report the first example of a FMS for all of the stereoisomers of the biologically active peptide tenuecyclamide  $A^{9c}$  (1) showcasing the f-Fmoc reagents as tools for effective peptide isomer library synthesis.

# ■ RESULTS AND DISCUSSION

Tenuecyclamides A (1a) and B (1b) are cyclic hexapeptides that differ only in configuration at a single stereocenter, and they were both isolated from the cyanobacterium Nostoc spongiaeforme var. tenue in 1998 (Figure 1). Although the exact stereochemistry was not elucidated at the time of isolation and structural identification, the assignme[nt](#page-1-0) [of](#page-7-0) the configuration was subsequently achieved by Kelly's group in the work that employed solid-phase peptide synthesis.<sup>10</sup> Tenuecyclamide A (1a) was found to inhibit the division of sea urchin embryos with an  $ED_{100}$  of 10.8  $\mu$ M. To date, th[e b](#page-8-0)iological activity of tenuecyclamide B  $(1b)$  has not been explored in detail.<sup>9c</sup>

The first step in the synthesis of the tenuecyclamide family was to protect L-alanine  $(2a)$  and D-alanine  $(2b)$  usi[ng](#page-8-0) two different f-Fmoc reagents bearing  $C_4F_9$  tags and  $C_6F_{13}$  tags,

Received: July 29, 2013 Published: September 16, 2013

**ACS** Publications

<span id="page-1-0"></span>

Figure 1. Structure of tenuecyclamide A (1a) and B (1b).

respectively (Scheme 1). The  $f_4$ -Fmoc reagent means bis-C<sub>4</sub>F<sub>9</sub> tagged Fmoc, and the  $f_6$ -Fmoc reagent means bis- $C_6F_{13}$  tagged

Scheme 1. N-Terminal Protection of Both L- and D-Alanine by f-Fmoc Reagents



Fmoc reagent in all of the schemes. In this process, 1,4-dioxane was employed for the  $f_6$ -Fmoc protection simply because it afforded greater solubility for the fluorous reagent in comparison to the more commonly used MeCN.

Subsequently, using the synthesis reported by Kelly, $10$  we prepared the two isomers of the thiazole-ring unit (5a and 5b) as a mixture, where the two isomers are encoded by a diff[e](#page-8-0)rent fluorous tag (Scheme 2). To rapidly purify the reaction





a Reagents and conditions: (a) 2.0 equiv of allyl alcohol, 1.1 equiv of HOBt, 1.1 equiv of HBTU, 2.1 equiv of DIEA, DMF, rt, 24 h, quant; (b) excess DEA, MeCN; (c) 1.0 equiv each of  $f_4$ -Fmoc-L- and  $f_6$ -Fmoc-D-alanine, 2.2 equiv of HOBt, 2.2 equiv of HBTU, 4.2 equiv of DIEA, DMF, rt, 12 h, 95% (two steps); FSPE, 55% aq THF on fluorous silica gel; (d) 18.0 equiv of Ph<sub>3</sub>PO, 9.0 equiv of Tf<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 2 h, 90%; FSPE, 70% aq MeOH on fluorous silica gel; (e) activated  $MnO<sub>2</sub>$ ,  $(10 \text{ w/w})$ , CHCl<sub>3</sub>, reflux, 30 min, 75%.

mixtures and remove organic reagents and byproducts from the desired fluorous tagged thiazoles after the reaction, FSPE was conducted in steps "c" and "d". In practice, the condensation reaction in step "c" was accomplished by using HBTU and HOBt. At the end of the reaction, the crude mixture was loaded onto fluorous silica gel<sup>11</sup> with MeOH, and this was followed by an initial elution with aqueous MeOH which removed the nontagged organic c[om](#page-8-0)pounds (reagents and reagent byproducts) from the reaction mixture while the fluorous tagged components were retained. A second elution with THF washed the fluorous fraction off the column and as a result, the f-Fmoc compounds were isolated in high yields without a tedious purification procedure. The two components of 5a and 5b were then separated by preparative  $f$ -HPLC<sup>8</sup> using a fluorous HPLC

 $\text{column,}^{12}$  and the isolated individual compounds were subsequently deprotected to give the enantiomerically pure thiazoles [6](#page-8-0)a and 6b (Scheme 3).





<sup>a</sup> Preparative f-HPLC conditions: 0 to 30 min 80% MeCN/H<sub>2</sub>O up to 100% MeCN. Flow rate: 7.0 mL/min; 254 nm; 6a, 80%; 6b, 80%.

Similarly, a mixture of the oxazole-ring unit (7-mix) was prepared from the tagged mixture of L- and D-alanine synthesized previously. Scheme 4 shows the synthetic route

Scheme 4. Preparation of  $7$ -mix<sup>a</sup>

3a and 3b\n
$$
3a \text{ and } 3b
$$
\n
$$
3b \text{ and } 3b
$$
\n
$$
a, b, c \text{ (FSPE)}, d
$$
\n
$$
a \text{ and } 3b
$$

a Reagents and conditions: (a) 2.2 equiv of L-threonine-OBn, 4.7 equiv of HOBt, 4.7 equiv of HBTU, 4.2 equiv of DIEA, DMF, rt, 24 h, quant; (b) 8.0 equiv of Dess-Martin periodinane, CH<sub>2</sub>Cl<sub>2</sub>, reflux, 2 h, 98%; (c) 18.0 equiv of Ph<sub>3</sub>PO, 9.0 equiv of Tf<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 2 h, quant; FSPE, 55% aq THF on fluorous silica gel; (d) H<sub>2</sub>, 20 mol % Pd/C, MeOH/THF, rt, 2 h, 94%.

to access 7-mix as a mixture of two compounds bearing either a  $C_4F_9$  or a  $C_6F_{13}$  chain. In this case, FSPE was conducted after step "c" to readily obtain the desired oxazole mixture 7-mix.

The synthesis then proceeded by splitting the 7-mix oxazole mixture into two parts, with one part being condensed with 6a and the other half with 6b, to give 8a-mix and 8b-mix, respectively, each as a mixture of two compounds that are orthogonally tagged (Scheme 5). 8a-mix and 8b-mix were then each divided again into two parts to give a total of four compound mixtures (Scheme 6). The allyl protecting groups in all four mixtures were then deprotected separately to reveal the corresponding acid and e[ach](#page-2-0) mixture was subsequently condensed again with either 6a or 6b, respectively. The mixtures were purified by FSPE, and as a result, we obtained

# Scheme 5. Preparation of 8a-mix and 8b-mix $^a$



<sup>a</sup>Reagents and conditions: (a) 2.4 equiv of  $6a$ , 4.7 equiv of HOBt, 4.7 equiv of HBTU, 4.2 equiv of DIEA, DMF, rt, 12 h, 91%; FSPE, 55% aq THF on fluorous silica gel; (b) 2.4 equiv of 6b, 4.7 equiv of HOBt, 4.7 equiv of HBTU, 4.2 equiv of DIEA, DMF, rt, 12 h, 98%; FSPE, 55% aq THF on fluorous silica gel.

<span id="page-2-0"></span>

<sup>a</sup>Reagents and conditions: (a) 20 mol %  $Pd(PPh<sub>3</sub>)<sub>4</sub>$ , 2.0 equiv of PhSiH<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt, 30 min; (b) 2.4 equiv of 6a, 4.7 equiv of HOBt, 4.7 equiv of HBTU, 4.2 equiv of DIEA, DMF, rt, 12 h; FSPE, 50% aq THF on fluorous silica gel; 9bc-mix, 85%; 9ef-mix, 87% (two steps); (c) 2.4 equiv of 6b, 4.7 equiv of HOBt, 4.7 equiv of HBTU, 4.2 equiv of DIEA, DMF, rt, 12 h; FSPE, 50% aq THF on fluorous silica gel; 9ad-mix, 80%; 9gh-mix, 82% (two steps). <sup>b</sup>Preparative f-HPLC conditions: 0−7.5 min 80% MeCN/H<sub>2</sub>O up to 100% MeCN. Flow rate: 7.0 mL/min.

four pairs of compounds, namely 9bc-mix, 9ad-mix, 9ef-mix, and 9gh-mix, each containing two compounds with different, and thus chromatographically separable, fluorine content.

As a demonstration, we analyzed the 9bc-mix by f-HPLC and confirmed that the  $C_4F_9$ -Fmoc compound 9b appeared at 10.8 min, whereas the  $C_6F_{13}$ -Fmoc compound 9c appeared at 23.8 min, a separation of 13.0 min completely dictated by the relative fluorine content of each molecule (Figure 2). With such



Figure 2. f-HPLC analysis of 9bc-mix. HPLC conditions: FluoroFlash HPLC column, 4.6 mm i.d., 150 mm length; 0 to 7.5 min 80% MeCN/ H2O up to 100% MeCN. Flow rate: 1.0 mL/min; 254 nm.

remarkable differences in retention time, each component of the four two component mixtures in Scheme 6, 9bc-mix, 9admix, 9ef-mix and 9gh-mix, were separated to the corresponding individual compounds 9a−h in almost quantitative yield using preparative f-HPLC (Scheme 6).

After exploring a variety of conditions to effect the key macrolactamization, using fully deprotected 9b as the test substrate, $13$  we identified that employing a fluorous tagged Mukaiyama reagent<sup>14</sup>  $10^{15}$  was superior to other plausible coupling [re](#page-8-0)agents. Further reaction optimization revealed that highly dilute reactio[n c](#page-8-0)on[dit](#page-8-0)ions, and syringe-pump addition of the substrate were required to obtain superior yields. The highest yield of 71% (three-step yield) for the macrolactamization was obtained when a DMF solution of 10 was added dropwise over 24 h to a 0.5 mM DMF solution of 9b. At the end of the cyclization reaction, the byproduct fluorouspyridone from 10 and other reagents used in the deprotection process were effectively separated from the desired material by precipitation. To achieve this, water was added to the reaction mixture after the coupling, and the resulting suspension was filtered. After a short column chromatography, further purification of the products was not necessary as they were found to be pure by <sup>1</sup>H NMR. Having optimized the lactamization conditions, we were poised to complete the synthesis of all eight stereoisomers of tenuecyclamide A. This was accomplished by treating compounds 9a−h individually with the fluorous coupling reagent 10 after the f-Fmoc groups had been removed with DEA and the allyl groups had been removed using palladium catalysis. In this manner we synthesized all eight target lactams (1a−g) with yields ranging from 41 to 71% over the three step lactamization and deprotection sequence (Scheme 7). The <sup>1</sup>H NMR of both tenuecyclamide  $A$  (1a) and  $B$  (1b) obtained here were completely consistent with the [dat](#page-3-0)a reported by Kelly.<sup>10</sup> In addition, <sup>1</sup>H NMR of both of 1e and 1g, were entirely consistent with the report.

## ■ CONCLUSION

In summary, we have conducted a concise synthesis of all eight stereoisomers of tenuecyclamide A (1a) based on an encoding strategy of the amino acid stereocenters using f-Fmoc reagents. If all stereoisomers were synthesized individually by the same linear synthetic route as outlined in this work, 136 synthetic steps would have been required; however, we carried out the syntheses in a mere 52 steps which includes the f-Fmoc protections. We believe that the f-Fmoc strategy will be a useful methods for divergent polypeptide synthesis. Additional studies and the examination of the biological activity of the novel

<span id="page-3-0"></span>Scheme 7. Intramolecular Cyclization of  $9a-h^a$ 



a<br>Reagents and conditions: (a) excess DEA, MeCN, rt, 30 min; (b) 10 mol % Pd(OAc)<sub>2</sub>, 0.8 equiv of PS-PPh<sub>3</sub>, 2.0 equiv of PhSiH<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt, 30 min; (c) 5.0−20.0 equiv of fluorous tagged Mukaiyama reagent 10, DMAP, TEA, DMF, 50 °C, 24 h, high dilution conditions; 1a, 48%; 1b, 71%; 1c, 70%; 1d, 51%; 1e, 41%; 1f, 49%; 1g, 43%; 1h, 48% (three steps).

unnatural isomers of tenuecyclamide A obtained by this work (1c−h) will be reported in due course.

## **EXPERIMENTAL SECTION**

General Experimental Methods. Except as otherwise indicated, all reactions were carried out under a positive pressure of nitrogen. All the laboratory chemicals were purchased and used without purification. Solvents were removed at a heating bath temperature of 40 °C and reduced pressure by rotary evaporation. Nonvolatile compounds were dried in vacuo at 0.01 mbar. All reactions were magnetically stirred and monitored by thin layer chromatography (TLC) using silica gel plates. Purification by chromatography was performed on silica gel (65−210 Å). NMR spectra were recorded at 270 MHz ( $^{1}$ H), 67.8 MHz ( $^{13}$ C), and 466 MHz ( $^{19}$ F), respectively. Chemical shifts  $\delta$  are referred in terms of ppm, and *J-coupling* constants are given in Hz. Abbreviations for multiplicity are as follows: s (singlet), d (doublet), t (triplet), q (quadruplet), quin (quintet signal), m (multiplet), br (broad signal). HPLC analysis and separations were performed on FluoroFlash columns (20 mm i.d., 250 mm length and 4.6 mm i.d., 150 mm length), and the oligopeptide analogue were purified by HPLC with UV/vis detector. High-resolution mass spectra were measured by FAB and EI using a magnetic sector analyzer. Highresolution FAB and EI mass spectra were calibrated with Ultramark 1621 and PFK, respectively, prior to data acquisition. Optical rotations were measured at a Na D-line ( $\lambda = 589$  nm) using a 100 mm cell.

 $f_4$ -Fmoc-L-alanine 3a. A solution of  $f_4$ -Fmoc-OSu (1.20 g, 1.45 mmol) in MeCN (20 mL) was added to a solution of L-alanine (193 mg, 2.17 mmol) in 10% aq  $\text{Na}_2\text{CO}_3$  (5 mL) slowly at 0 °C. After 5 min, the reaction mixture was warmed to room temperature and stirred for 3 h. The reaction acidified with 1.0 M HCl aq to pH 2.0. The organic layer was extracted with diethyl ether, dried over  $\text{Na}_2\text{SO}_4$ , and concentrated. The residue was purified by column chromatography on silica gel (CHCl<sub>3</sub>/MeOH = 50:1) to give  $f_4$ -Fmoc-L-alanine (1.11 g, 95%): white solid; mp 75–76 °C;  $[\alpha]_D^{25} = +1.0$  (c = 0.10, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.67 (d, J = 7.3 Hz, 2H), 7.42 (s, 2H), 7.25 (d, J = 7.8 Hz, 2H), 5.27 (d, J = 7.3 Hz, 1H), 4.44−4.39 (m, 3H), 4.18 (t, J = 6.9 Hz, 1H), 3.02−2.96 (m, 4H), 2.51−2.32 (m, 4H), 1.49 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$  177.9, 155.8, 144.6, 144.3, 139.7 (2C), 138.1 (2C), 127.9 (2C), 125.0 (2C), 124.9 (2C), 120.1−111.0 (8C), 67.0, 49.5, 47.0, 26.5 (4C), 18.5; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.87 (4F), -124.27 (4F), -114.66 (4F), -80.91 (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{30}H_{23}F_{18}NO_4$ 803.1340, found 803.1418.

 $f_6$ -Fmoc- $D$ -alanine 3b.  $f_6$ -Fmoc- $D$ -alanine was synthesized from  $f_6$ -Fmoc-OSu and D-alanine in 96% yield as a white solid by following the procedure used for the synthesis of  $f_4$ -**Fmoc-L-alanine 3a**: white solid; mp 112−113 °C;  $[\alpha]_D^{25} = -1.9$  (c = 0.10, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.67 (d, J = 7.3 Hz, 2H), 7.42 (s, 2H), 7.25 (d, J = 7.3 Hz, 2H), 5.26 (d, J = 7.8 Hz, 1H), 4.48–4.37 (m, 3H), 4.19 (t, J = 6.6 Hz, 1H), 3.02−2.96 (m, 4H), 2.52−2.32 (m, 4H), 1.49 (d, J = 7.0 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$  177.8, 155.9, 144.6, 144.3, 139.7 (2C), 138.1 (2C), 127.9 (2C), 125.0 (2C), 124.9, 124.8, 120.1− 106.6 (12C), 67.0, 49.5, 47.0, 26.6 (2C), 26.5 (2C), 18.1; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -126.06 (4F), -123.35 (4F), -122.78 (4F), −121.79 (4F), −114.46 (4F), −80.72 (6F); HRMS [FAB+] m/z calcd for C<sub>34</sub>H<sub>23</sub>F<sub>26</sub>NO<sub>4</sub> 1003.1212, found 1003.1248.

Fmoc-Cys-S-Tr-OAllyl.<sup>10b</sup> HOBt $\cdot$ H<sub>2</sub>O (288 mg, 1.88 mmol) and HBTU (712 mg, 1.88 mmol) were added to a solution of N-Fmoc-Strityl-L-cysteine (1.00 g, [1.](#page-8-0)71 mmol) in DMF (10 mL) at room temperature. After 15 min, allyl alcohol (232  $\mu$ L, 3.41 mmol) and DIEA (614  $\mu$ L, 3.76 mmol) were added separately, and the resulting mixture was stirred at the same temperature for 24 h. The reaction mixture was diluted with ethyl acetate and washed with  $H_2O$ , saturated aq NaHCO<sub>3</sub>, and brine, and the organic layer was dried over  $Na<sub>2</sub>SO<sub>4</sub>$ and then concentrated. The crude residue was purified by silica gel chromatography (EtOAc/hexane =1:3) to give the title compound (1.07 g, quant): white solid; mp 70–72 °C; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.76 (d, J = 7.3 Hz, 2H), 7.60 (d, J = 7.0 Hz, 2H), 7.41–7.38 (m, 8H), 7.33−7.18 (m, 11H), 5.95−5.80 (m, 1H), 5.33−5.23 (m, 3H), 4.62 (d, J = 5.1 Hz, 2H), 4.38−4.35 (m, 3H), 4.23 (t, J = 7.0 Hz, 1H), 2.68- 2.64 (m, 2H).

 $f_4/f_6$ -Fmoc-Ala-Cys-S-Tr-OAllyl. To a solution of N-Fmoc-S-trityl-Lcysteine-OAllyl (171 mg, 0.274 mmol) in MeCN (4 mL) was added DEA (4 mL) at room temperature. The mixture was stirred at the same temperature for 30 min. After concentration in vacuo, the reaction mixture was azeotroped to dryness with MeCN  $(3 \times 3 \text{ mL})$ . In another flask, to a solution mixture of  $f_4$ -Fmoc-L-alanine (100 mg, 0.124 mmol) and  $f_6$ -Fmoc-D-alanine (125 mg, 0.124 mol) in DMF (4) mL) were added  $HOBt·H<sub>2</sub>O$  (42 mg, 0.274 mmol) and HBTU (104 mg, 0.274 mol) at room temperature. After 15 min, a solution of the above residue in DMF  $(2 \text{ mL})$  and DIEA  $(88 \mu L, 0.523 \text{ mmol})$  was added. The reaction mixture was stirred at room temperature for 12 h. After the addition of 1.0 M HCl aq and then dilution with ethyl acetate, the organic layer was washed with  $H_2O$ , saturated aq Na $HCO_3$ and brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated. The crude residue was purified by FSPE. A column was packed with fluorous silica gel (4.98 g) using 55% aq THF as the solvent. The crude reaction mixture was then loaded onto this column and eluted with 43 mL of 55% aq THF followed by 65 mL of THF. Evaporation of the THF fraction by vacuum centrifuge gave the title product (304 mg, 95%): pale yellow solid; mp 68–74 °C; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.67 (d, J = 8.1) Hz, 4H), 7.42−7.37 (m, 16H), 7.28−7.15 (m, 22H), 6.45 (d, J = 5.4 Hz, 1H), 6.16 (d, J = 7.8 Hz, 1H), 5.94−5.21 (m, 2H), 5.44−5.21 (m, 6H), 4.59−4.51 (m, 6H), 4.41−4.35 (m, 6H), 4.16 (t, J = 6.5 Hz, 2H), 3.00−2.95 (m, 8H), 2.78−2.58 (m, 4H), 2.49−2.33 (m, 8H), 1.49− 1.26 (m, 6H); <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  –126.00 (4F), –125.85 (4F), −124.23 (4F), −123.29 (4F), −122.72 (4F), −121.75 (4F), −114.65 (4F), −114.38 (4F), −80.92 (6F), −80.66 (6F).

 $f_{\frac{1}{2}}/f_{6}$ -Fmoc-thiazoline Ring Unit. To a solution of triphenylphosphine oxide (388 mg, 1.40 mmol) in dry  $CH_2Cl_2$  (3 mL) was added trifluoromethanesulfonic anhydride (0.11 mL, 0.698 mmol) slowly at 0 °C under N<sub>2</sub>. After the reaction mixture was stirred at 0 °C for 10 min, a solution of  $f_4/f_6$ -Fmoc-alanine-S-trityl-L-cysteine-OAllyl (400 mg, 0.155 mmol) in dry  $CH_2Cl_2$  (3 mL) was added. The reaction mixture was stirred at 0 °C for 2 h. The reaction was quenched with saturated aq NaHCO<sub>3</sub>. The organic layer was washed with  $H<sub>2</sub>O$ , dried over Na2SO4, and concentrated. The crude residue was purified by FSPE. A column was packed with fluorous silica gel (22.1 g) using 70% aq MeOH as the solvent. The crude reaction mixture was then loaded onto this column and eluted with 120 mL of 70% aq MeOH followed by 80 mL of THF. The evaporation of the THF fraction by vacuum centrifuge gave the title product (287 mg, 90%): white solid; mp 68− 72 °C; <sup>I</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.68 (d, J = 8.1 Hz, 4H), 7.44 (s, 4H), 4.32−7.23 (m, 4H), 5.99−5.85 (m, 2H), 5.64 (d, J = 6.8 Hz, 1H), 5.55 (d, J = 7.0 Hz, 1H), 5.39- 5.29 (m, 2H), 5.27−5.23 (m, 2H), 5.18−5.10 (m, 2H), 4.69−4.68 (m, 6H), 4.40 (m, 4H), 4.19 (t, J = 7.0 Hz, 2H), 3.63−3.55 (m, 4H), 3.02−2.96 (m, 8H), 2.52−2.35 (m, 8H), 1.49 (d, J = 4.6 Hz, 6H); <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.99

(4F), −125.83 (4F), −124.22 (4F), −123.29 (4F), −122.71 (4F), −121.71 (4F), −114.66 (4F), −114.43 (4F), −80.89 (6F), −80.62 (6F).

Compound 5a and 5b. A solution of ring-closing product (198 mg, 0.0962 mmol) and activated MnO<sub>2</sub> (1982 mg, 10 w/w) in dry CHCl<sub>3</sub> (10 mL) was stirred at 60 °C for 30 min under N<sub>2</sub>. After MnO<sub>2</sub> was removed by filtration through a pad of Celite and washed with CHCl<sub>3</sub>, the filtrate was concentrated. The crude residue was purified by silica gel chromatography (EtOAc/hexane =  $1/2$ ) to give 5a and 5b (148) mg, 75%): white solid; mp 94−104 °C; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.09 (s, 2H), 7.66 (d, J = 7.3 Hz, 4H), 7.43 (s, 4H), 7.23 (d, J = 7.8 Hz, 4H), 6.10−5.95 (m, 2H), 5.67 (d, J = 5.9 Hz, 2H), 5.40 (dd, J = 17.0, 1.4 Hz, 2H), 5.29 (dd, J = 10.5, 1.4 Hz, 2H), 5.19 (quin, J = 6.6 Hz, 2H), 4.84 (d, J = 5.9 Hz, 4H), 4.52–4.37 (m, 4H), 4.16 (t, J = 5.5 Hz, 2H), 3.01−2.95 (m, 8H), 2.51−2.32 (m, 8H), 1.65 (d, J = 5.9 Hz, 6H); <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -126.00 (4F), -125.85 (4F), −124.22 (4F), −123.26 (4F), −122.69 (4F), −122.0 (4F), −114.73 (4F), −114.47 (4F), −80.88 (6F), −80.63 (6F). Fluorous analytical HPLC (flow rate; 1.0 mL/min; 0 to 30 min: 80% MeCN to 100% MeCN; 30 to 60 min: 100% MeCN):  $t_R = 22.0$  min ( $f_4$ -Fmoc-(S)thiazole), 41.9 min ( $f_6$ -Fmoc-(R)-thiazole).

Preparative Separation of 5a and 5b. The preparative separation of 5a and 5b was carried out on a SHIMADZU HPLC system. Compounds 5a and 5b were dissolved in THF (2.0 mL) and filtered through a syringe filter (0.45  $\mu$ m pore size) prior to injection. The separation was carried out on a FluoroFlash HPLC column, 20 mm i.d., 250 mm length. The separation was achieved by gradient elution with 80%  $MeCN/H<sub>2</sub>O$  up to 100%  $MeCN$  in 30.0 min, followed by isocratic elution with 100% MeCN over 60 min with a constant flow rate of 7.0 mL/min. A UV detector (254 nm) was used to manually identify the peaks. 5a and 5b (40 mg) was injected chromatographic run. The yield of the demixing was 80%, and the following two compounds were isolated:  $f_4$ -Fmoc-(S)-thiazole-ring unit 5a (14 mg) and  $f_6$ -Fmoc- $(R)$ -thiazole-ring unit 5b (18 mg).

 $f_4$ -Fmoc-(S)-thiazole-ring unit 5a: white solid; mp 110−111 °C;  $[\alpha]_{\text{D}}^{25}$  = -5.64 (c = 0.13, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.10  $(s, 1H)$ , 7.68 (d, J = 7.8 Hz, 2H), 7.43 (s, 2H), 7.25 (d, J = 7.3 Hz, 2H), 6.11−5.96 (m, 1H), 5.55 (d, J = 5.1 Hz, 1H), 5.41 (dd, J = 17.3, 1.4 Hz, 1H), 5.30 (dd, J = 10.5, 1.4 Hz, 1H), 5.22−5.17 (m, 1H), 4.85 (d, J = 5.9 Hz, 2H), 4.52–4.39 (m, 2H), 4.18 (t, J = 6.2 Hz, 1H),  $3.02 - 2.96$  (m, 4H),  $2.51 - 2.32$  (m, 4H), 1.66 (d, J = 5.9 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>),  $\delta$  173.7, 160.9, 155.8, 147.0 (2C), 144.6 (2C), 144.3 (2C), 139.8 (2C), 138.1, 131.9 (2C), 128.0, 127.5, 125.1, <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.85 (4F), -124.21 (4F), −114.59 (4F), −80.86 (6F); HRMS [FAB+] m/z calcd for  $C_{36}H_{29}F_{18}N_2O_4S$  927.1562, found 927.1513.

 $f_6$ -Fmoc-(R)-thiazole-ring unit 5b: white solid; mp 75–76 °C;  $[\alpha]_D^{25}$  = +8.24 (c = 0.12, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.10  $($ s, 1H), 7.68 (d, J = 8.1 Hz, 2H), 7.43 (s, 2H), 7.25 (d, J = 7.8 Hz, 2H), 6.10−5.96 (m, 1H), 5.57 (d, J = 6.5 Hz, 1H), 5.41 (dd, J = 17.0, 1.4 Hz, 1H), 5.30 (dd, J = 10.5, 1.4 Hz, 1H), 5.22−5.17 (m, 1H), 4.85  $(d, J = 5.9 \text{ Hz}, 2H), 4.52-4.39 \text{ (m, 2H)}, 4.17 \text{ (d, } J = 6.6 \text{ Hz}, 1H),$ 3.02−2.96 (m, 4H), 2.51−2.39 (m, 4H), 1.66 (d, J = 5.9 Hz, 3H); 13C NMR (67.8 MHz, CDCl<sub>3</sub>), δ 173.6, 160.8, 155.5, 147.0 (2C), 144.5 (2C), 144.4 (2C), 139.7 (2C), 138.1, 131.8 (2C), 128.0, 127.4, 125.0, 124.9, 120.1−108.2 (13C), 66.9, 66.0, 49.3, 47.1, 26.6 (2C), 22.7 (2C), 21.1; <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.99 (4F), -123.28 (4F), −122.71 (4F), −121.70 (4F), −114.38 (4F), −80.66 (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{40}H_{29}F_{26}N_2O_4S$  1127.1435, found 1127.1387.

Compound 6a. Diethylamine  $(2 \text{ mL})$  was added to a solution of  $f_4$ -Fmoc-(S)-thiazole-ring unit 5a (10 mg, 0.0108 mmol) in MeCN (2 mL), and the reaction mixture was stirred at room temperature for 30 min. After concentration in vacuo, the reaction mixture was azeotroped to dryness with MeCN  $(3 \times 3 \text{ mL})$ . The crude residue was purified by FSPE. A column was packed with fluorous silica gel (0.32 g) using 50% aq THF as the solvent. The crude reaction mixture was then loaded onto this column and eluted with 2 mL of 50% aq THF followed by 4 mL of THF. The evaporation of the THF fraction by vacuum centrifuge gave 6a as colorless oil, which was used without purification (2.3 mg, quant): colorless oil;  $\left[\alpha\right]_{\text{D}}^{25} = -13.3$  ( $c = 0.265$ , CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (s, 1H), 6.11–5.97 (m, 1H), 5.41 (dd,  $J = 17.0$  Hz, 1.4 Hz, 1H), 5.30 (dd,  $J = 10.5$  Hz, 1.4 Hz, 1H), 4.86 (dd, J = 5.9 Hz, 1.4 Hz, 2H), 4.50 (q, J = 6.8 Hz, 1H), 1.86 (s, 2H), 1.55 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$ 179.9, 161.1, 146.6, 131.9, 127.4, 118.8, 65.9, 49.7, 24.8; HRMS [FAB +]  $m/z$  calcd for C<sub>9</sub>H<sub>13</sub>N<sub>2</sub>O<sub>2</sub>S 213.0698, found 213.0709.

Compound 6b. Compound 6b was prepared from  $f_6$ -Fmoc- $(R)$ thiazole-ring unit 5b in quantitative yield as colorless oil by following the procedure used for prepare of 6a: colorless oil;  $[\alpha]_{D}^{25} = +9.86$  ( $c =$ 0.191, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (s, 1H), 6.12– 5.97 (m, 1H), 5.41 (dd,  $J = 17.3$  Hz, 1.4 Hz, 1H), 5.29 (dd,  $J = 10.5$ Hz, 1.4 Hz, 1H), 4.85 (dd,  $J = 5.9$  Hz, 1.4 Hz, 2H), 4.50 (q,  $J = 6.8$  Hz, 1H), 1.86 (s, 2H), 1.55 (d, J = 6.5 Hz, 3H); 13C NMR (67.8 MHz, CDCl3) δ 179.9, 161.1, 146.6, 131.9, 127.4, 118.8, 65.9, 49.7, 24.8; HRMS [FAB+]  $m/z$  calcd for calcd for  $C_9H_{13}N_2O_2S$  213.0698, found 213.0688.

 $f_4/f_6$ -Fmoc-Ala-Thr-OBn. A solution of  $f_4$ -Fmoc-L-alanine (6.00 g, 7.47 mmol) and  $f_6$ -Fmoc-D-alanine (7.47 g, 7.47 mmol) in DMF (120 mL) was treated with HOBt·H2O (5.37 g, 35.10 mmol) and HBTU (13.31 g, 35.10 mmol) at room temperature. After 15 min, H-Thr-OBn·(COOH)<sub>2</sub> (4.92 g, 16.43 mmol) and DIEA (5.3 mL, 31.36 mmol) were added to the above reaction mixture. The reaction mixture was stirred at room temperature for 24 h. After the addition of 1.0 M HCl aq and then dilution with ethyl acetate, the organic layer was washed with H<sub>2</sub>O, saturated aq NaHCO<sub>3</sub>, and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The crude residue was pass through a short pad of silica gel (EtOAc/hexane =1:3) to obtain the title compounds (15.80 g, 94%): pale yellow solid; mp 115−118 °C; <sup>1</sup> H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.67 (d, J = 7.3 Hz, 4H), 7.42 (s, 4H), 7.32–7.23 (m, 16H), 6.80−6.65 (m, 2H), 5.47−5.42 (m, 2H), 5.23−5.12 (m, 4H), 4.64 (dd, J = 8.4, 1.9 Hz, 2H), 4.40–4.38 (m, 6H), 4.16 (t, J = 6.9 Hz, 2H), 3.02−2.96 (m, 8H), 2.52−2.32 (m, 8H), 2.18−2.02 (m, 2H), 1.46−1.39 (m, 6H), 1.21−1.18 (m, 6H); 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  −126.04 (4F), −125.89 (4F), −124.25 (4F), −123.30 (4F), −122.74 (4F), −121.77 (4F), −114.69 (4F), −114.46 (4F), −80.89  $(6F)$ ,  $-80.65$   $(6F)$ .

 $f_4/f_6$ -Fmoc-Ala-Thr(C=O)-OBn. Dess-Martin periodinane (155 mg, 0.0365 mol) was added to a solution of  $f_4/f_6$ -Fmoc-dipeptides (100 mg, 0.0457 mmol) in  $CH_2Cl_2$  (4 mL). The reaction mixture was stirred at 40 °C for 2 h. The reaction was quenched with saturated aq sodium hyposulfite/saturated aq NaHCO<sub>3</sub> =  $1/1$  v/v. The organic layer was washed with saturated aq NaHCO<sub>3</sub>, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The crude residue was passed through a short pad of silica gel to obtain the title compounds (99 mg, 98%): pale yellow solid; mp 86–88 °C; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.68 (d, J = 7.3 Hz, 4H), 7.42 (s, 4H), 7.34−7.33 (m, 12H), 7.25 (d, J = 7.6 Hz, 4H), 5.33−5.15 (m, 8H), 4.44−4.39 (m, 6H), 4.18 (t, J = 6.2 Hz, 2H), 3.03−2.96 (m, 8H), 2.52−2.30 (m, 14H), 1.41 (t, J = 7.3 Hz, 6H); <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -126.02 (4F), -125.88 (4F), -124.24 (4F), −123.29 (4F), −122.73 (4F), −121.74 (4F), −114.70 (4F), −114.42 (4F), −80.87 (6F), −80.65 (6F).

 $f_{\frac{1}{6}}$ -Fmoc-oxazole Ring Unit. To a solution of triphenylphosphine oxide (18.33 g, 65.89 mmol) in dry  $CH_2Cl_2$  (120 mL) was added trifluoromethanesulfonic anhydride (5.1 mL, 32.95 mmol) slowly at 0  $^{\circ}$ C under N<sub>2</sub>. After the reaction mixture was stirred at same temperature for 10 min, a solution of  $f_4/f_6$ -Fmoc-oxidative products  $(8.00 \text{ g}, 3.66 \text{ mmol})$  in dry  $\text{CH}_2\text{Cl}_2$   $(60 \text{ mL})$  was added to the above reaction mixture. The reaction mixture was stirred at 0 °C for 2 h. The reaction was quenched with saturated aq NaHCO<sub>3</sub>. The organic layer was washed with H<sub>2</sub>O, dried over  $\text{Na}_2\text{SO}_4$ , and concentrated. The crude residue was purified by FSPE. A column was packed with fluorous silica gel (405.0 g) using 55% aq THF as the solvent. The crude reaction mixture was then loaded onto this column and eluted with 1020 mL of 55% aq THF followed by 1120 mL of THF. The evaporation of the THF fraction by vacuum centrifuge gave the title compounds (7.95 g, quant): white solid; mp 81−84 °C; <sup>1</sup> H NMR  $(270 \text{ MHz}, \text{CDCl}_3) \delta$  7.67 (d, J = 7.8 Hz, 4H), 7.44–7.23 (m, 18H), 5.53 (d,  $J = 7.8$  Hz, 2H), 5.36 (s, 4H), 5.00 (quin,  $J = 6.4$  Hz, 2H), 4.42 (d, J = 6.5 Hz, 4H), 4.16 (t, J = 6.9 Hz, 2H), 3.02–2.96 (m, 8H), 2.58 (s, 6H), 2.51−2.32 (m, 8H), 1.56−1.55 (m, 6H); 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  –126.01 (4F), –125.87 (4F), –124.25 (4F), –123.32 (4F), −122.72 (4F), −121.75 (4F), −114.67 (4F), −114.47 (4F),  $-80.87$  (6F),  $-80.63$  (6F).

Compound 7-mix. The ring closing product  $(f_4/f_6\text{-}F$ moc-oxazolering unit)  $(1.00 \text{ g}, 0.465 \text{ mmol})$  and  $\overline{Pd}/C$   $(11 \text{ mg}, 0.0930 \text{ mmol})$  in THF/MeOH (10 mL,  $1/1$  v/v) was stirred at room temperature for 2 h under  $H_2$  balloon. The catalyst was removed by filtration through a pad of Celite and washed with ethyl acetate. The filtrate was concentrated to give 7-mix (865 mg, 94%). The residue was immediately used for the next step without further purification: colorless oil; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.66 (d, J = 7.8 Hz, 4H), 7.44−7.41 (m, 4H), 7.23 (d, J = 7.0 Hz, 4H), 5.07−5.02 (m, 2H), 4.40  $(d, J = 6.5$  Hz, 4H), 4.15  $(t, J = 6.6$  Hz, 2H), 3.00–2.94  $(m, 8H)$ , 2.62 (s, 6H), 2.51−2.25 (m, 8H), 1.58 (d, J = 7.3 Hz, 6H); 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -126.01 (4F), -125.86 (4F), -124.24 (4F), -123.29 (4F), −122.73 (4F), −121.74 (4F), −114.63 (4F), −114.41 (4F),  $-80.91$  (6F),  $-80.65$  (6F).

Compound 8a-mix. A solution of 7-mix (250 mg, 0.127 mmol) in DMF (4 mL) was treated with HOBt·H2O (91 mg, 0.597 mmol) and HBTU (226 mg, 0.597 mmol) at room temperature. After 15 min, a solution of  $6a$  (67 mg, 0.305 mmol) in DMF  $(4 mL)$  and DIEA  $(91$  $\mu$ L, 0.55 mmol) were sequentially added to the above solution. The reaction mixture was stirred at room temperature for 12 h. After the addition of 1.0 M HCl aq and dilution with ethyl acetate, the organic layer was washed with  $H_2O$ , saturated aq Na $HCO_3$ , and brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated. The crude residue was purified by FSPE. A short column was packed with fluorous silica gel (5.1 g) using 55% aq THF as the solvent. The crude reaction mixture was then loaded onto this column and eluted with 62.9 mL of 55% aq THF followed by 25.0 mL of THF. The evaporation of the THF fraction by vacuum centrifuge gave product 8a-mix (272 mg, 91%): white solid; mp 123−126 °C; <sup>1</sup> H NMR (270 MHz, CDCl3) δ 8.10−8.07 (m, 2H), 7.70−7.67 (m, 4H), 7.47−7.43 (m, 6H), 7.26−7.24 (m, 4H), 6.09− 5.95 (m, 2H), 5.57−5.52 (m, 2H), 5.43−5.27 (m, 6H), 5.05−4.94 (m, 2H), 4.85−4.82 (m, 4H), 4.46−4.44 (m, 4H), 4.21 (t, J = 6.3 Hz, 2H), 3.01−2.96 (m, 8H), 2.62 (s, 6H), 2.51−2.32 (m, 8H), 1.75−1.71 (m, 6H), 1.62−1.53 (m, 6H); <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  –125.99 (4F), −125.86 (4F), −124.25 (4F), −123.32 (4F), −122.72 (4F), −121.73 (4F), −114.64 (4F), −114.42 (4F), −80.87 (6F), −80.65 (6F).

Compound 8b-mix. Compound 8b-mix was synthesized from 7 mix and 6b in 98% yield as a white solid by following the procedure used for the synthesis of 8a-mix: white solid; mp 121−124 °C; <sup>1</sup> H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.10–8.07 (m, 2H), 7.68 (d, J = 8.1 Hz, 4H), 7.46−7.43 (m, 6H), 7.25 (d, J = 6.8 Hz, 4H), 6.10−5.94 (m, 2H), 5.54 (quin, J = 7.1 Hz, 2H), 5.43−5.26 (m, 6H), 4.99 (quin, J = 6.4 Hz, 2H), 4.85–4.81 (m, 4H), 4.46–4.44 (m, 4H), 4.20 (t,  $J = 6.6$ Hz, 2H), 3.02−2.96 (m, 8H), 2.61 (s, 6H), 2.50−2.31 (m, 8H), 1.73  $(d, J = 6.8 \text{ Hz}, 6\text{H})$ , 1.53  $(d, J = 7.8 \text{ Hz}, 6\text{H})$ ; <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -126.04 (4F), -125.85 (4F), -124.24 (4F), -123.29 (4F), −122.69 (4F), −121.71 (4F), −114.62 (4F), −114.39 (4F), −80.88  $(6F)$ ,  $-80.64$   $(6F)$ .

Compound 9bc-mix. A solution of 8a-mix (210 mg, 0.0891 mmol) and  $Pd(PPh<sub>3</sub>)<sub>4</sub>$  (21 mg, 0.0178 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was stirred at room temperature for 10 min in  $N_2$ . To the solution mixture was added PhSiH<sub>3</sub> (22  $\mu$ L, 0.178 mmol), and then the mixture was stirred at room temperature for 30 min. The mixture was through a pad of Celite and washed  $CH_2Cl_2$ . The filtrate was concentrated. The residue was immediately used for the next step without further purification: A solution of the above residue in DMF (6 mL) was treated with HOBt $\cdot$ H<sub>2</sub>O (64 mg, 0.419 mmol) and HBTU (159 mg, 0.419 mmol) at room temperature. After 15 min, a solution of 6a (45 mg, 0.214 mmol) in DMF  $(2 \text{ mL})$  and DIEA  $(63 \mu L, 0.374 \text{ mmol})$ were sequentially added to the above solution at room temperature. The mixture was stirred at the same temperature for 12 h. After the addition of 1.0 M HCl aq and dilution with ethyl acetate, the organic layer was washed with  $H_2O$ , saturated aq NaHCO<sub>3</sub>, and brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated. The crude residue was purified by FSPE. A short column was packed with fluorous silica gel (7.2 g) using 50% aq THF as the solvent. The crude reaction mixture was then loaded onto this column and eluted with 120.0 mL of 50% aq THF followed by 25.0 mL of THF. The evaporation of the THF fraction by vacuum centrifuge gave product 9bc-mix as a colorless semisolid (202 mg, 85%). Fluorous analytical HPLC (flow rate; 1.0 mL/min; 0 to 7.5 min: 80% MeCN/H2O to 100% MeCN; 7.5 to 30 min: 100% MeCN):  $t_{\text{R}}$  =10.8 min (9b), 23.8 min (9c).

Compound 9ad-mix. Compound 9ad-mix was synthesized from 8a-mix and 6b in 80% yield as a colorless semisolid by following the procedure used for the synthesis of 9bc-mix. Fluorous analytical HPLC (flow rate; 1.0 mL/min; 0 to 7.5 min: 80% MeCN/H<sub>2</sub>O to 100% MeCN; 7.5 to 30 min: 100% MeCN):  $t_R = 14.9$  min (9a), 23.8 min (9d).

Compound 9ef-mix. Compound 9ef-mix was synthesized from 8bmix and 6a in 87% yield as a colorless semisolid by following the procedure used for the synthesis of 9bc-mix. Fluorous analytical HPLC (flow rate; 1.0 mL/min; 0 to 7.5 min: 80% MeCN/H<sub>2</sub>O to 100% MeCN; 7.5 to 30 min: 100% MeCN):  $t_R = 15.5$  min (9e), 23.1 min (9f).

Compound 9gh-mix. Compound 9gh-mix was synthesized from 8b-mix and 6b in 82% yield as a colorless semisolid by following the procedure used for the synthesis of 9bc-mix. Fluorous analytical HPLC (flow rate; 1.0 mL/min; 0 to 7.5 min: 80% MeCN/ $H_2O$  to 100% MeCN; 7.5 to 30 min: 100% MeCN):  $t_R = 15.1$  min (9g), 23.1 min (9h).

Preparative Separation of 9bc-mix. Compound 9bc-mix was dissolved in THF (1.5 mL) and filtered through a syringe filter (0.45  $\mu$ m pore size) prior to injection. The separation was carried out on a FluoroFlash HPLC column, 20 mm i.d., 250 mm length. The separation was achieved by gradient elution with 80%  $MeCN/H<sub>2</sub>O$  up to 100% MeCN in 7.5 min, followed by isocratic elution with 100% MeCN over 60 min with a constant flow rate of 7.0 mL/min. A UV detector (254 nm) was used to manually identify the peaks. 9bc-mix (78 mg) was injected for the chromatographic run. The yield of the demixing was quantitative yield and the following two compounds were isolated:  $9b(30 \text{ mg})$  and  $9c(48 \text{ mg})$ .

Compound 9b: colorless semisolid;  $[\alpha]_{D}^{25} = -11.2$  (c = 0.2, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.10 (s, 1H), 8.00 (s, 1H), 7.87 (d, J = 8.6 Hz, 1H), 7.67 (d, J = 8.1 Hz, 2H), 7.43 (d, J = 6.5 Hz, 2H), 7.32 (d, J = 7.8 Hz, 1H), 7.25 (d, J = 8.6 Hz, 2H), 6.09−5.94 (m, 1H), 5.59 (quin, J = 7.2 Hz, 1H), 5.50 (quin, J = 7.3 Hz, 1H), 5.42− 5.26 (m, 3H), 4.99 (quin,  $J = 5.9$  Hz, 1H), 4.83 (d,  $J = 5.9$  Hz, 2H),  $4.45-4.43$  (m, 2H),  $4.18$  (t, J = 6.5 Hz, 1H),  $3.01-2.95$  (m, 4H),  $2.62$  $(s, 3H)$ , 2.51–2.31 (m, 4H), 1.79 (d, J = 7.3 Hz, 3H), 1.70 (d, J = 6.8) Hz, 3H), 1.57 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$ 173.2, 173.0, 161.3, 161.2, 161.1, 160.9, 160.8, 160.4 (2C), 155.4, 154.2, 149.0, 146.7 (2C), 144.6, 144.2, 139.6 (2C), 138.0 (2C), 131.8 (2C), 128.4 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.1−110.4 (9C), 66.9, 65.9, 47.1 (2C), 46.7, 45.2, 26.5 (2C), 20.9 (3C), 11.6; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.85 (4F), -124.27 (4F), -114.61 (4F),  $-80.85$  (6F); HRMS [FAB+]  $m/z$  calcd for C<sub>49</sub>H<sub>43</sub>F<sub>18</sub>N<sub>6</sub>O<sub>7</sub>S<sub>2</sub> 1233.2347, found 1233.2366.

**Compound 9c:** colorless semisolid;  $[\alpha]_{D}^{25} = +3.63$  ( $c = 0.2$ , CHCl<sub>3</sub>);<br><sup>1</sup>H NMR (270 MHz CDCL)  $\delta$  8 10 ( $s$  1H) 8 00 ( $s$  1H) 7 87 (d I-<sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.10 (s, 1H), 8.00 (s, 1H), 7.87 (d, J = 8.6 Hz, 1H), 7.67 (d,  $J = 8.1$  Hz, 2H), 7.43 (d,  $J = 7.3$  Hz, 2H), 7.32  $(d, J = 8.6 \text{ Hz}, 1H)$ , 7.24  $(d, J = 8.1 \text{ Hz}, 2H)$ , 6.09–5.94  $(m, 1H)$ , 5.59 (quin, J = 7.4 Hz, 1H), 5.50 (quin, J = 7.4 Hz, 1H), 5.42−5.26 (m, 3H), 4.99 (quin, J = 6.5 Hz, 1H), 4.83 (d, J = 5.9 Hz, 2H), 4.47−4.43  $(m, 2H)$ , 4.18 (t, J = 6.5 Hz, 1H), 3.01–2.95  $(m, 4H)$ , 2.62 (s, 3H), 2.50−2.31 (m, 4H), 1.79 (d, J = 7.3 Hz, 3H), 1.70 (d, J = 6.8 Hz, 3H), 1.57 (m, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$  173.2, 173.0, 161.4, 161.3, 161.1, 160.9, 160.4 (2C), 155.4, 154.2, 149.0, 146.7 (2C), 144.6, 144.2, 139.7 (2C), 138.1 (2C), 131.8 (2C), 128.5 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.2−108.2 (13C), 67.1, 65.9, 47.1, 46.7, 46.6, 45.2, 26.6 (2C), 21.0 (2C), 20.9, 11.6; <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$ −125.98 (4F), −123.31 (4F), −122.66 (4F), −121.68 (4F), −114.41 (4F), -80.61 (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{53}H_{43}F_{26}N_6O_7S_2$ 1433.2221, found 1433.2236.

Preparative Separation of 9ad-mix. The preparative separation of 9ad-mix was carried out in the same manner as 9bc-mix. Compound 9ad-mix (223 mg) was injected for the chromatographic run. The yield of the demixing was 96%, and the following two compounds were isolated: 9a (102 mg) and 9d (113 mg).

Compound 9a: colorless semisolid;  $\left[\alpha\right]_D^{25} = -14.4$  (c = 0.088, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.11 (s, 1H), 8.01 (s, 1H), 7.90 (d, J = 7.8 Hz, 1H), 7.67 (d, J = 7.8 Hz, 2H), 7.43 (d, J = 7.3 Hz, 2H), 7.35 (d, J = 7.8 Hz, 1H), 7.24 (d, J = 7.8 Hz, 2H), 6.10−5.94 (m, 1H), 5.59 (quin, J = 7.7 Hz, 1H), 5.50 (quin, J = 7.6 Hz, 1H), 5.43− 5.26 (m, 3H), 6.07–5.97 (m, 1H), 4.83 (d, J = 9.5 Hz, 2H), 4.51–4.39  $(m, 2H)$ , 4.18 (t, J = 6.2 Hz, 1H), 3.00–2.94  $(m, 4H)$ , 2.61 (s, 3H), 2.50−2.31 (m, 4H), 1.78 (d, J = 7.3 Hz, 3H), 1.70 (d, J = 6.5 Hz, 3H), 1.58 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$  173.1, 173.0, 161.3, 161.2, 161.1, 160.9, 160.4 (2C), 155.4, 154.2, 149.0, 146.7 (2C), 144.6, 144.3, 139.7, 139.6, 138.1 (2C), 131.8 (2C), 128.4 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.0−110.9 (9C), 66.9, 65.9, 47.0, 46.7, 46.6, 45.2, 26.6 (2C), 26.5, 20.9 (2C), 11.6; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.85 (4F), -124.24 (4F), -114.63 (4F), -80.89 (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{49}H_{43}F_{18}N_6O_7S_2$  1233.2347, found 1233.2344.

Compound 9d: colorless semisolid;  $[\alpha]_{\text{D}}^{25} = +6.92$  (c = 0.095, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (s, 1H), 8.01 (s, 1H), 7.86 (d, J = 8.1 Hz, 1H), 7.68 (d, J = 7.8 Hz, 2H), 7.43 (d, J = 6.8 Hz, 2H), 7.33 (d, 8.4 Hz, 1H), 7.25 (d, J = 8.1 Hz, 2H), 6.10−5.96 (m, 1H), 5.59 (quin, J = 7.3 Hz, 1H), 5.50 (quin, J = 7.3 Hz, 1H), 5.44− 5.27 (m, 3H), 5.00 (quin,  $J = 6.9$  Hz, 1H), 4.85 (d,  $J = 5.9$  Hz, 2H),  $4.52-4.40$  (m, 2H),  $4.18$  (t, J = 6.6 Hz, 1H),  $3.01-2.95$  (m, 4H), 2.63 (s, 3H), 2.51−2.30 (m, 4H), 1.79 (d, J = 7.3 Hz, 3H), 1.71 (d, J = 7.3 Hz, 3H), 1.57 (d, J = 7.3 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$ 173.1, 173.0, 161.3, 161.2, 161.1, 160.9, 160.4 (2C), 155.4, 154.2, 149.0, 146.7 (2C), 144.6, 144.3, 139.7 (2C), 138.1 (2C), 131.8, 131.7, 128.5 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.2−108.5 (13C), 66.9, 65.9, 47.1, 46.7, 46.6, 45.2, 26.6 (2C), 25.6, 20.9 (2C), 11.6; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.98 (4F), -123.30 (4F), -122.70 (4F), −121.72 (4F), −114.39 (4F), −80.64 (6F); HRMS [FAB+] m/z calcd for  $C_{53}H_{43}F_{26}N_6O_7S_2$  1433.2221, found 1433.2207.

Preparative Separation of 9ef-mix. The preparative separation of 9ef-mix was carried out in the same manner as for 9bc-mix. Compound 9ef-mix (90 mg) was injected for the chromatographic run. The yield of the demixing was 98%, and the following two compounds were isolated: **9e**  $(42 \text{ mg})$  and **9f**  $(46 \text{ mg})$ .

Compound 9e: colorless semisolid;  $\left[\alpha\right]_{\text{D}}^{25} = -14.9$  (c = 0.06, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (s, 1H), 8.04 (s, 1H), 7.88 (d, J = 7.8 Hz, 1H), 7.68 (d, J = 7.8 Hz, 2H), 7.43 (s, 2H), 7.32  $(d, J = 7.8 \text{ Hz}, 1H), 7.25 (d, J = 7.8 \text{ Hz}, 2H), 6.08-5.97 (m, 1H), 5.59$ (quin, J = 7.3 Hz, 1H), 5.50 (quin, J = 7.3 Hz, 1H), 5.43−5.26 (m, 3H), 4.99 (quin, J = 5.9 Hz, 1H), 4.84 (d, J = 4.6 Hz, 2H), 4.53−4.39 (m, 2H), 4.19 (t, J = 6.6 Hz, 1H), 3.00−2.95 (m, 4H), 2.62 (s, 3H), 2.50−2.34 (m, 4H), 1.78 (d, J = 6.5 Hz, 3H), 1.70 (d, J = 7.0 Hz, 3H), 1.56 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$  173.2, 173.1, 161.4, 161.2 (2C), 160.9, 160.4 (2C), 155.5, 154.3, 149.1, 146.9, 146.8, 144.6, 144.4, 139.7 (2C), 138.1 (2C), 131.9 (2C), 128.5 (2C), 128.0, 127.8 (2C), 125.0, 124.1, 120.2−110.4 (9C), 67.0, 66.0, 49.7, 46.8, 46.7, 45.2, 26.6 (2C), 21.0 (3C), 11.5; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  −125.88 (4F), −124.23 (4F), −114.58 (4F), −80.86 (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{49}H_{43}F_{18}N_6O_7S_2$  1233.2347, found 1233.2397.

**Compound 9f:** colorless semisolid;  $[\alpha]_D^{25} = +13.1$  (c = 0.12, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (s, 1H), 8.01 (s, 1H), 7.87 (d, J = 7.8 Hz, 1H), 7.68 (d, J = 7.3 Hz, 2H), 7.43 (d, J = 6.8 Hz, 2H), 7.33 (d, J = 7.8 Hz, 1H), 7.25 (d, J = 7.3 Hz, 2H), 6.10−5.94 (m, 1H), 5.59 (quin, J = 7.2 Hz, 1H), 5.50 (quin, J = 7.3 Hz, 1H), 5.44− 5.27 (m, 3H), 5.00 (quin,  $J = 6.8$  Hz, 1H), 4.84 (d,  $J = 5.1$  Hz, 2H), 4.48−4.44 (m, 2H), 4.18 (t, J = 6.5 Hz, 1H), 3.01−2.95 (m, 4H), 2.62  $(s, 3H)$ , 2.51–2.32 (m, 4H), 1.79 (d, J = 6.5 Hz, 3H), 1.71 (d, J = 7.3 Hz, 3H), 1.58 (d, J = 8.1 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$ 173.2, 173.1, 161.4, 161.1, 161.0 (2C), 160.4 (2C), 155.9, 155.3, 149.0, 146.7 (2C), 144.6, 144.3, 139.7, 139.6, 138.1 (2C), 131.8 (2C), 128.4 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.0−108.4 (13C), 66.9, 65.9,

47.0, 46.7, 46.6, 45.1, 26.6 (2C), 21.0 (2C), 20.9, 11.6; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  –126.09 (4F), –123.35 (4F), –122.77 (4F), –121.78 (4F),  $-114.42$  (4F),  $-80.75$  (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{53}H_{43}F_{26}N_6O_7S_2$  1433.2221, found 1433.2268.

Preparative Separation of 9gh-mix. The preparative separation of 9gh-mix was carried out in the same manner as for 9bc-mix. Compound 9gh-mix (140 mg) was injected for the chromatographic run. The yield of the demixing was 97%, and the following two compounds were isolated: 9g (64 mg) and 9h (72 mg).

Compound 9g: colorless semisolid;  $[\alpha]_{\text{D}}^{25} = -1.57$  (c = 0.05, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.11 (s, 1H), 8.04 (s, 1H), 7.95 (d, J = 7.8 Hz, 1H), 7.68 (d, J = 7.3 Hz, 2H), 7.39−7.26 (m, 3H), 7.25 (d, J = 8.6 Hz, 2H), 6.09−5.93 (m, 1H), 5.60 (quin, J = 7.6 Hz, 1H), 5.50 (quin, J = 7.3 Hz, 1H), 5.43−5.26 (m, 3H), 4.99 (quin, J = 7.1 Hz, 1H), 4.83 (d, J = 5.4 Hz, 2H), 4.52−4.39 (m, 2H), 4.19 (t, J = 6.2 Hz, 1H), 3.00−2.95 (m, 4H), 2.62 (s, 3H), 2.50−2.31 (m, 4H), 1.76 (d, J = 6.4 Hz, 3H), 1.70 (d, J = 7.0 Hz, 3H), 1.56 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>) δ 173.2, 173.1, 161.3, 161.2, 161.0, 160.9, 160.4 (2C), 155.4, 154.2, 149.0, 146.7 (2C), 144.6, 144.3, 139.7 (2C), 138.1 (2C), 131.8 (2C), 128.4 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.2−110.0 (9C), 66.9, 65.9, 47.1 (2C), 46.7, 45.2, 26.6, 26.5, 20.9 (3C), 11.6; <sup>19</sup>F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  -125.84 (4F), −124.24 (4F), −114.62 (4F), −80.86 (6F); HRMS [FAB+] m/z calcd for C<sub>49</sub>H<sub>43</sub>F<sub>18</sub>N<sub>6</sub>O<sub>7</sub>S<sub>2</sub> 1233.2347, found 1233.2377.

Compound **9h**: colorless semisolid;  $[\alpha]_D^{25} = +12.4$  (c = 0.087, CHCl<sub>3</sub>); <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.10 (s, 1H), 8.00 (s, 1H), 7.88 (d, J = 8.6 Hz, 1H), 7.67 (d, J = 7.8 Hz, 2H), 7.43 (d, J = 6.8 Hz, 2H), 7.33 (d, J = 7.8 Hz, 1H), 7.25 (d, J = 7.8 Hz, 2H), 6.09−5.94 (m, 1H), 5.59 (quin, J = 7.3 Hz, 1H), 5.50 (quin, J = 7.3 Hz, 1H), 5.43− 5.26 (m, 3H), 5.00 (quin,  $J = 7.4$  Hz, 1H), 4.83 (d,  $J = 6.5$  Hz, 2H),  $4.46-4.44$  (m, 2H),  $4.18$  (t,  $J = 6.6$  Hz, 1H),  $3.01-2.95$  (m, 4H),  $2.62$  $(s, 3H)$ , 2.51–2.32 (m, 4H), 1.79 (d, J = 7.3 Hz, 3H), 1.70 (d, J = 6.8) Hz, 3H), 1.57 (d, J = 7.0 Hz, 3H); <sup>13</sup>C NMR (67.8 MHz, CDCl<sub>3</sub>)  $\delta$ 173.2, 173.1, 161.2 (2C), 161.1, 160.9, 160.4 (2C), 155.4, 154.2, 149.0, 146.7 (2C), 144.6, 1443, 139.7, 139.6, 138.1(2C), 131.8 (2C), 128.4 (2C), 127.9, 127.7 (2C), 124.9, 124.0, 120.2−107.1 (13C), 66.9, 65.9, 47.1, 46.7, 45.2, 43.4, 26.7 (2C), 21.0 (2C), 20.9, 11.6; 19F NMR (466 MHz, CDCl<sub>3</sub>)  $\delta$  –125.97 (4F), –123.28 (4F), –122.71 (4F), –121.58 (4F),  $-114.45$  (4F),  $-80.62$  (6F); HRMS [FAB+]  $m/z$  calcd for  $C_{53}H_{43}F_{26}N_6O_7S_2$  1433.2221, found 1433.2218.

Procedure for Cyclization Using  $9a$ . Diethylamine  $(2 \text{ mL})$  was added to a solution of  $9a$  (28 mg, 0.0230 mmol) in MeCN (2 mL), and the mixture was stirred at rt for 30 min. After concentration in vacuo, the reaction mixture was azeotroped to dryness with MeCN (3  $\times$  3 mL). Pd(OAc)<sub>2</sub> (0.5 mg, 0.00230 mmol) and styrene polymerbound triphenylphosphene (11 mg, 1.7 mol/g, 0.0184 mmol) were added to a flask containing dry  $CH_2Cl_2$  (3 mL) under N<sub>2</sub>. After the mixture was stirred for 15 min, a solution of the above residue in dry  $CH_2Cl_2$  (2 mL) and PhSiH<sub>3</sub> (6  $\mu$ L, 0.0461 mmol) were added separately. The reaction progress was monitored by TLC, and the reaction was complete in 30 min. The reaction mixture was passed through a pad of Celite and concentrated. The residue was used in the next step without further purification. The resulting amino acid derivative was dissolved in DMF (16 mL) solution. This solution was added to a flask containing fluorous tagged Mukaiyama reagent (373 mg, 0.461 mmol), TEA (192  $\mu$ L, 1.38 mmol), and DMAP (23 mg, 0.184 mmol) in dry DMF (32 mL) at 50 °C over 8 h using a syringe pump. After the addition was complete, the mixture was stirred at the same temperature for 24 h. To the reaction mixture was added  $H_2O$ (12 mL) and then filtered. The filtrate was diluted with ethyl acetate and washed with 1.0 M HCl aq,  $H<sub>2</sub>O$ , and brine. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The crude residue was purified by short silica gel chromatography (EtOAc/hexane =  $4/1$ ) to obtain 1a as a colorless semisolid (5.1 mg, 48%).

Tenucyclamide A (**1a**):<sup>10à</sup> colorless semisolid; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.69–8.68 (m, 2H), 8.61 (d, J = 6.8 Hz, 1H), 8.24 (s, 1H), 8.17 (s, 1H), 5.60 (quin,  $J = 6.6$  $J = 6.6$  $J = 6.6$  Hz, 1 H), 5.37 (quin,  $J = 6.4$  Hz, 1H), 5.14 (quin,  $J = 6.3$  Hz, 1H), 2.70 (s, 3H), 1.80 (d,  $J = 6.8$  Hz, 3H), 1.71−1.69 (m, 6H).

<span id="page-7-0"></span>Tenuecyclamide B (1b).<sup>10a</sup> Compound 1b was synthesized from 9b in 71% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: [colo](#page-8-0)rless semisolid;  $^{1}H$  NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.70–8.66 (m, 2H), 8.60 (d, J = 7.8 Hz, 1H), 8.18 (s, 1H), 8.14 (s, 1H), 5.68 (quin, J = 7.0 Hz, 1H), 5.44 (quin, J = 6.6 Hz, 1H), 5.24 (quin, J = 6.9 Hz, 1H), 2.69 (s, 3H), 1.73−1.71 (m, 9H).

Compound 1c. Compound 1c was synthesized from 9c in 70% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: colorless semisolid;  $[\alpha]_{D}^{25} = -24.0$  (c = 0.21, CHCl<sub>3</sub>);<br><sup>1</sup>H NMB (270 MHz, CDCl)  $\delta$  8.70 (d, I – 6.5 Hz, 1H) 8.65–8.60 <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.70 (d, J = 6.5 Hz, 1H), 8.65–8.60  $(m, 2H)$ , 8.20 (s, 1H), 8.17 (s, 1H), 5.58 (quin, J = 6.6 Hz, 1H), 5.42  $(quin, J = 6.6 Hz, 1H), 5.21 (quin, J = 6.6 Hz, 1H), 2.69 (s, 3H), 1.79$ (d, J = 6.8 Hz, 3H), 1.70−1.65 (m, 6H); 13C NMR (67.8 MHz, CDCl3) δ 171.7, 170.6, 161.3, 160.7, 159.7, 159.6, 154.2, 149.2, 148.3, 128.6, 124.6, 123.1, 48.0, 47.9, 44.2, 24.7, 24.5, 19.9, 11.7; HRMS [EI +]  $m/z$  calcd for C<sub>19</sub>H<sub>20</sub>N<sub>6</sub>O<sub>4</sub>S<sub>2</sub> 460.0987, found 460.0953.

Compound 1d. Compound 1d was synthesized from 9d in 51% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: colorless semisolid;  $[\alpha]_{D}^{25} = +40.3$  ( $c = 0.29$ , CHCl<sub>3</sub>);<br><sup>1</sup>H NMB (270 MHz, CDCl)  $\delta$  8.68–8.66 (m, 2H) 8.56 (d, I – 6.8 <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.68–8.66 (m, 2H), 8.56 (d, J = 6.8 Hz, 1H), 8.21 (s, 1H), 8.12 (s, 1H), 5.56 (quin,  $J = 6.5$  Hz, 1H), 5.41 (quin,  $J = 6.6$  Hz, 1H),  $5.28$  (quin,  $J = 6.8$  Hz, 1H), 2.70 (s, 3H), 1.72−1.70 (m, 9H); <sup>13</sup>C NMR (67.8 Hz, CDCl<sub>3</sub>) δ 171.7, 170.6, 161.3, 160.7, 159.7, 159.6, 154.2, 149.2, 148.3, 128.6, 124.6, 123.1, 48.0, 47.9, 44.2, 24.7, 24.5, 19.9, 11.7; HRMS [EI+] m/z calcd for  $C_{19}H_{20}N_6O_4S_2$  460.0987, found 460.0953.

Compound 1e.<sup>10a</sup> Compound 1e was synthesized from 9e in 41% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: [color](#page-8-0)less semisolid; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$ 8.68−8.66 (m, 2H), 8.57 (d, J = 5.9 Hz, 1H), 8.21 (s, 1H), 8.12 (s, 1H), 5.56 (quin,  $J = 6.8$  Hz, 1H), 5.41 (quin,  $J = 6.4$  Hz, 1H), 5.28 (quin,  $J = 6.9$  Hz, 1H), 2.70 (s, 3H), 1.72–1.68 (m, 9H).

Compound 1f. Compound 1f was synthesized from 9f in 49% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: colorless semisolid;  $[\alpha]_{D}^{25} = +2.4$  ( $c = 0.144$ , CHCl<sub>3</sub>);<br><sup>1</sup>H NMR (270 MHz, CDCl)  $\delta$  8.60–8.67 (m, 2H) 8.61 (d, I – 6.5) <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.69–8.67 (m, 2H), 8.61 (d, J = 6.5 Hz, 1H), 8.24 (s, 1H), 8.17 (s, 1H), 5.60 (quin,  $J = 6.8$  Hz, 1H), 5.36  $(quin, J = 6.3 Hz, 1H), 5.14 (quin, J = 6.1 Hz, 1H), 2.70 (s, 3H), 1.79$  $(d, J = 6.8 \text{ Hz}, 3\text{H})$ , 1.73–1.64 (m, 6H); <sup>13</sup>C NMR (67.8 Hz, CDCl<sub>3</sub>) δ 172.2, 171.4, 161.9, 160.7, 160.0, 159.6, 153.6, 148.8, 148.6, 128.6, 124.9, 123.9, 48.3, 48.0, 44.9, 24.9, 23.7, 20.8, 11.6; HRMS [EI+] m/z calcd for  $C_{19}H_{20}N_6O_4S_2$  460.0987, found 460.0980.

Compound  $1g^{10a}$  Compound 1g was synthesized from 9g in 43% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: c[olor](#page-8-0)less semisolid; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$ 8.70 (d, J = 5.9 Hz, 1H), 8.65−8.60 (m, 2H), 8.20 (s, 1H), 8.17 (s, 1H), 5.58 (quin,  $J = 6.6$  Hz, 1H), 5.42 (quin,  $J = 6.6$  Hz, 1H), 5.21 (quin, J = 6.6 Hz, 1H), 2.69 (s, 3H), 1.79 (d, J = 6.8 Hz, 3H), 1.70− 1.65 (m, 6H).

Compound 1h. Compound 1h was synthesized from 9h in 48% yield as a colorless semisolid by following the procedure used for the synthesis of 1a: colorless semisolid;  $[\alpha]_{D}^{25} = +4.14$  ( $c = 0.98$ , CHCl<sub>3</sub>);<br><sup>1</sup>H NMR (270 MHz, CDCl)  $\delta$  8.70–8.66 (m, 2H) 8.59 (d, I – 7.8) <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  8.70–8.66 (m, 2H), 8.59 (d, J = 7.8 Hz, 1H), 8.18 (s, 1H), 8.13 (s, 1H), 5.67 (quin, J = 6.9 Hz, 1H), 5.44 (quin,  $J = 6.6$  Hz, 1H), 5.24 (quin,  $J = 6.8$  Hz, 1H), 2.69 (s, 3H), 1.73−1.68 (m, 9H); <sup>13</sup>C NMR (67.8 Hz, CDCl<sub>3</sub>) δ 171.4, 170.9, 161.6, 160.5, 159.7, 159.4, 153.9, 148.9, 148.5, 128.4, 124.3, 123.5, 47.5, 47.3, 44.2, 25.0, 24.9, 20.8, 11.6; HRMS [EI+] m/z calcd for  $C_{19}H_{20}N_6O_4S_2$  460.0987, found 460.0941.

#### ■ ASSOCIATED CONTENT

#### **S** Supporting Information

Proton, carbon, and fluorine NMR data for all new compounds and fluorous HPLC chart of 5a, 5b, 9bc-mix, 9ad-mix, 9ef-mix, and 9gh-mix. This material is available free of charge via the Internet at http://pubs.acs.org.

# ■ AUTHOR INFORMATION

#### Corresponding Author

\*E-mail: matsugi@meijo-u.ac.jp.

#### Notes

The auth[ors declare no compet](mailto:matsugi@meijo-u.ac.jp)ing financial interest.

#### ■ ACKNOWLEDGMENTS

This work was partially supported by the Ministry of Education, Science, Sports and Culture, Grant-in-aid for Scientific Research (C) 23580154 and the Research Institute of Meijo University. We thank Meiji Seika Pharma Co., Ltd. for funding this work.

#### ■ **DEDICATION**

We dedicate this paper to Prof. Dennis P. Curran on the occasion of his 60th birthday.

#### ■ REFERENCES

(1) (a) Curran, D. P.; Hadida, S.; He, M. J. Org. Chem. 1997, 62, 6714. (b) Zhang, W.; Curran, D. P. Tetrahedron 2006, 62, 11837.

(2) Curran, D. P. Light Fluorous Chemistry-A Users Guide. In The Handbook of Fluorous Chemistry; Gladysz, J., Horváth, I., Curran, D. P., Eds.; Wiley-VCH: Weinheim, 2004; p 128.

(3) Curran, D. P. Separations with Fluorous Silica Gel and Related Materials. In The Handbook of Fluorous Chemistry; Gladysz, J., Horváth, I., Curran, D. P., Eds.; Wiley-VCH: Weinheim, 2004; p 101.

(4) For representative examples, see: (a) Luo, Z.; Zhang, Q.; Oderaotoshi, Y.; Curran, D. P. Science 2001, 291, 1766. (b) Zhang, Q.; Rivkin, A.; Curran, D. P. J. Am. Chem. Soc. 2002, 124, 5774. (c) Wei, Z. Tetrahedron 2003, 59, 4475. (d) Yang, F.; Newsome, J. J.; Curran, D. P. J. Am. Chem. Soc. 2006, 128, 14200. (e) Jung, W. H.; Guyenne, G.; Riesco-Fagundo, C.; Mancuso, J.; Nakamura, S.; Curran, D. P. Angew. Chem., Int. Ed. 2008, 47, 1130. (f) Zhang, W. Fluorine in Medicinal Chemistry and Chemical Biology; Ojima, I., Ed.; Wiley-Blackwell: Hoboken, 2009; p 335. (g) Zhang, K.; Gudipati, V.; Curran, D. P. Synlett 2010, 667. (h) Sui, B.; Yeh, E. A.-H.; Curran, D. P. J. Org. Chem. 2010, 75, 2942. (i) Bajpai, R.; Curran, D. P. J. Am. Chem. Soc. 2011, 133, 20435. (j) Moretti, J. D.; Wang, X.; Curran, D. P. J. Am. Chem. Soc. 2012, 134, 7963. (k) Curran, D. P.; Sinha, M. K.; Zhang, K.; Sabatini, J. J.; Cho, D.-H. Nature Chem. 2012, 4, 124. (l) Bajpai, R.; Curran, D. P. J. Am. Chem. Soc. 2011, 133, 20435.

(5) For representative examples, see: (a) Wipf, P.; Fritch, P. C.; Geib, S. J.; Sefler, A. M. J. Am. Chem. Soc. 1998, 120, 4105. (b) Sakurai, K.; Chung, H. K.; Kahne, D. J. Am. Chem. Soc. 2004, 126, 16288. (c) Lu, Q.; Wang, D. S.; Chen, C. S.; Hu, Y. D.; Chen, C. S. J. Med. Chem. 2005, 48, 5530. (d) Mroczkiewicz, M.; Winkler, K.; Nowis, D.; Placha, G.; Golab, J.; Ostaszewski, R. J. Med. Chem. 2010, 53, 1509. (e) Thornburg, C. C.; Thimmaiah, M.; Shaala, L. A.; Hau, A. M.; Malmo, J. M.; ssIshmael, J. E.; Youssef, D. T. A.; McPhail, K. L. J. Nat. Prod. 2011, 74, 1677.

(6) (a) Najera, C. ́ Synlett 2002, 1388. (b) Wuts, P. G. M.; Greene, T. W. In Green's Protective Groups in Organic Synthesis, 4th ed.; Wuts, P. G. M., Greene, T. W., Eds.; Wiley-Interscience: Hoboken, 2007; p 711. (7) (a) Matsugi, M.; Yamanaka, K.; Inomata, I.; Takekoshi, N.; Hasegawa, M.; Curran, D. P. QSAR Comb. Sci. 2006, 25, 713. (b) Sugiyama, Y.; Ishihara, K.; Masuda, Y.; Kobayashi, Y.; Hamamoto, H.; Matsugi, M. Tetrahedron Lett. 2013, 54, 2060.

(8) For use of f-HPLC in separation of fluorous compounds, see: (a) Curran, D. P.; Oderaotoshi, Y. Tetrahedron 2001, 57, 5243. (b) Curran, D. P.; Dandapani, S.; Werner, S.; Matsugi, M. Synlett 2004, 1545. (c) Zhang, W.; Lu, Y.; Chen, C. H.; Zeng, L.; Cassel, D. B. J. Comb. Chem. 2006, 8, 687. (d) Fukui, Y.; Burucner, A. M.; Shin, Y.; Balachandran, R.; Day, B. W.; Curran, D. P. Org. Lett. 2006, 8, 301.

(9) (a) Decikachary, T. V. In Cyanophyta; Decikachary, T. V., Ed.; Indian Council of Agricultural Research: New Delhi, 1956; p 372. (b) Ikegami, S.; Kawada, K.; Kimura, Y.; Suzuki, A. Agric. Biol. Chem. <span id="page-8-0"></span>1979, 43, 161. (c) Banker, R.; Carmle, S. J. Nat. Prod. 1998, 61, 1248. (d) Donia, M. S.; Ravel, J.; Schmidt, E. W. Nature Chem. Biol. 2008, 4, 341.

(10) (a) You, S.; Deechongkit, S.; Kelly, J. W. Org. Lett. 2004, 6, 2627. (b) You, S.; Kelly, J. W. J. Org. Chem. 2003, 68, 9506.

(11) Fluorous silica gel (FluoroFlash silica gel, 40  $\mu$ m), which is available from SIGMA-Aldrich, was used for all FSPE process.

(12) Preparative and analytical f-HPLC were conducted by FluoroFlash columns (20 mm i.d., 250 mm length and 4.6 mm i.d., 150 mm length) purchased from Fluorous Technologies, Inc. The corporation is closing; however, fluorous column (Wakopak Fluofix-II 120E) having almost same separation ability is available from Wako Pure Chemical Industries, Ltd.

(13) Although other condensation reagents such as PyBOP ((benzotriazol-1-yloxy) tripyrrolidinophosphonium hexafluorophosphate) or DPPA (diphenyl phosphorazidate) or DEPC (diethoxyphosphoryl cyanide) or Yamaguchi reagent (2,4,6-trichlorobenzoyl chloride) were examined under high dilution conditions (0.5 mM), fluorous tagged Mukaiyama reagent 8 was the most effective for the lactamization in terms of the yield and handling.

(14) Mukaiyama, T.; Usui, M.; Shimada, E.; Saigo, K. Chem. Lett. 1975, 1045.

(15) (a) Matsugi, M.; Nakamura, S.; Kunda, Y.; Sugiyama, Y.; Shioiri, T. Tetrahedron Lett. 2010, 51, 133. (b) Sugiyama, Y.; Kurata, Y.; Kunda, Y.; Miyazaki, A.; Matsui, J.; Nakamura, S.; Hamamoto, H.; Shioiri, T.; Matsugi, M. Tetrahedron 2012, 68, 3885.